Hirosaki Univ. Technology

New catalyst for inducing same oxidation reaction under illumination even in the dark

Can oxidize hydrazine in the dark

Overview

Photocatalysts are materials that induce redox reactions under illumination. In particular, it is effective for downhill reactions (oxidative degradation of hazardous and pollutants) and titanium oxide (TiO2) which is UV light responsive, is already in practical use. However, TiO2 photocatalytic technology is limited to small amounts and low concentrations of substances based on the solar light spectrum, and there is a problem that oxidation degradation treatment similar to that under illumination is never induced in the dark conditions.

In his research for application as a photoelectrode and photocatalyst for organic p-n junctions, the inventor found that organic p-n junctions can catalyze (= dual catalysis) the oxidation of thiols even in the dark conditions [1]. In addition, as a result of his intensive research, he succeeded in expanding the target of dual catalysis to formic acid, hydrogen peroxide, and hydrazine by supporting a cocatalyst on organic p-n junctions.

Dual catalysis is a new type of catalysis that TiO2 does not have. The selection of organic p-n junctions and co-catalysts is expected to expand the application range and market size of catalysts for environmental purification applications.

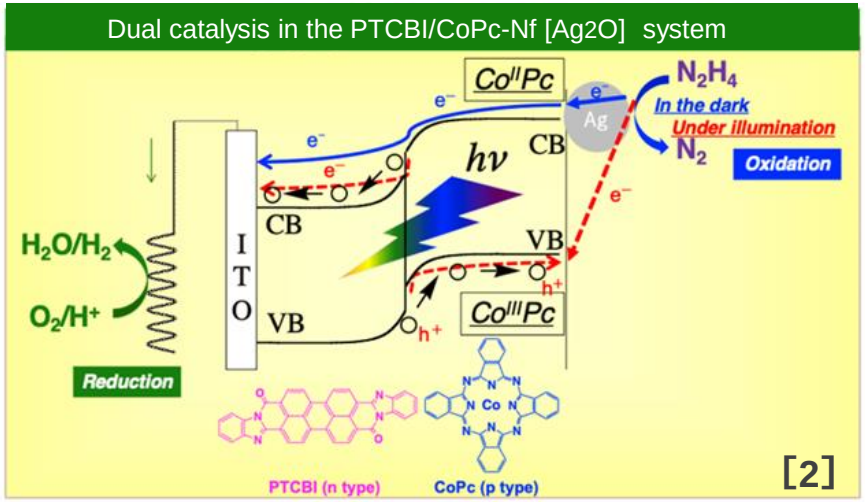
Product Application

- Catalysts for environmental purification
- Complementation of titanium oxide

IP Data

IP No.	:	JP2021-090760
Inventor	:	ABE Toshiyuki, SATO Mamoru
Admin No.	:	K23-027

Ag acts as a cocatalyst and oxidizes hydrazine even in the dark



Catalytic electrode		Amount of nitrogen generated (μL)		Amount of hydrogen generated (μL)	
		Under illumination	In the dark	Under illumination	In the dark
Example 1	ITO / PTCBI / CoPc + Ag_2O	48.4	8.04	94.8	14.3
comparative example 1	ITO / PTCBI / CoPc	46.0	0.0	78.5	4.21

Both under illumination and in the dark, hydrogen generation associated with the oxidation of hydrazine can be confirmed, and it is larger under illumination than in the dark.

= **Dual catalysis effect**

Related Works

[1] J. Mater. Chem. A, 2017, 5, 7445
[2] RSC Adv.,2020, 12, 1850

Contact

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